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Nickel-catalyzed highly chemoselective cocyclotrimerization of arynes with allenes: a novel method for 10-methylene-9,10-dihydrophenanthrenes[†]

Jen-Chieh Hsieh, Dinesh Kumar Rayabarapu and Chien-Hong Cheng*

Department of Chemistry, Tsing Hua University, Hsinchu 300, Taiwan. E-mail: chcheng@mx.nthu.edu.tw; Fax: 886-3-5724698; Tel: 886-3-5721454

Received (in Cambridge, UK) 8th December 2003, Accepted 22nd December 2003 First published as an Advance Article on the web 4th February 2004

The NiBr₂(dppe)–Zn system effectively catalyzes the [2+2+2] cocyclotrimerization of arynes with allenes, leading to 10-methylene-9,10-dihydrophenanthrenes in moderate to good yields. The cocyclotrimerization is highly selective with only the internal double bond of the allenes being involved in the reaction.

The transition-metal mediated [2+2+2] cocyclotrimerization of alkynes is a powerful method for the construction of polycyclic compounds in one step.¹ Although this cycloaddition has been known for decades, the control of both regio- and chemoselectivity remains a great challenge to organic chemists. Recently, the cyclotrimerization of arynes and alkynes has attracted great attention. Perez et al. revealed the cyclotrimerization of arynes catalyzed by palladium to give triphenylenes.² More recently, both the Perez and the Yamamoto groups reported the cyclotrimerization of benzynes and alkynes catalyzed by palladium complexes leading to the formation of phenanthrene derivatives.³ Our interest in nickel and cycloaddition chemistry^{4,5} encouraged us to explore the possibility of using allenes in the cocyclotrimerization with benzynes. Herein, we wish to report the first example of nickelcatalyzed cocyclotrimerization of two benzynes with an allene, providing an efficient method for the synthesis of 10-methylene-9,10-dihydrophenanthrene derivatives in a highly selective fashion (Scheme 1). Notably, the phenanthrene skeleton is found in a number of biologically active natural products.6

Treatment of benzyne precursor (1a) with cyclohexylallene (2a) in the presence of NiBr₂(dppe) (dppe: bis(diphenylphosphino)ethane), zinc powder and CsF in acetonitrile at 80 °C for 24 h afforded 9-cyclohexyl-10-methylene-9,10-dihydrophenanthrene (3a) in 67% yield. The structure of 3a that is derived from two molecules of 1a and one molecule of 2a was confirmed by its ¹H NMR, ¹³C NMR and mass data. Interestingly, the internal double bond of the allene participates in the cocyclotrimerization reaction and the product having an exocyclic C–C double bond does not undergo isomerization to give the more stable aromatic derivative. The reaction appears highly selective with respect to the double bond of the allene: the other isomer of 3a with the terminal double



† Electronic supplementary information (ESI) available: synthesis and characterization data of compounds 3. See http://www.rsc.org/suppdata/cc/ b3/b315795d/

bond of the allene being involved in the cocyclotrimerization was not observed.

To understand the nature of this nickel-catalyzed carbocyclization, the effect of solvent and nickel complex used on the reaction of **1a** with **2a** was investigated. In the absence of either nickel catalyst or zinc powder, no cyclotrimerization product **3a** was observed. Ni(PPh₃)₂Br₂ and Pd(PPh₃)₄ gave **3a** in 23% and 34% yields, respectively. A substantial amount of triphenylene resulting from homotrimerization of benzyne was observed. NiBr₂(dppe) was found to be the best catalyst, producing the desired **3a** in 67% yield and a trace of triphenylene. Solvent was also critical for this catalytic reaction. No reaction occurred in toluene, whereas THF afforded a trace of **3a**. CH₃CN was the best solvent when NiBr₂(dppe) was used as the catalyst.

The nickel-catalyzed cocyclotrimerization reaction was successfully extended to various substituted allenes and the results are listed in Table 1. Thus, 1a reacts with cyclopentyl- (2b) and cycloheptylallene (2c) to provide the corresponding cocyclotrimerization products 3b-c in 58 and 64% yields, respectively. Bulky tert-butylallene (2d) and long chain aliphatic allenes (2e-g) underwent [2+2+2] cycloaddition to produce 9,10-dihydrophenanthrene derivatives 3d-g in good yields (Scheme 1, Table 1, entries 4-7). A small amount of homo-trimerization product ([2+2+2]) of benzyne was observed in all these reactions. No reaction was observed for the cyclotrimerization of benzyne with phenylallene. For the reaction of dimethylallene 2h with 1a, two isomers 3h and 3h' in 1 : 1.9 molar ratio were observed with a combined yield of 67%. It appears that both C–C double bonds in **2h** are reactive in the reaction. The major isomer 3h' was produced with the terminal double bond of the allene participating in the cyclotrimerization reaction. Similarly, the reaction of 1a with disubstituted allene 2i proceeded to give isomers 3i and 3i' in a 2.9 : 1 ratio and total yield of 69%. In contrast to 3h', major isomer 3i in the present case is from the participation of the internal double bond of allene 2i. The present method can also be applied to the cocyclotrimerization of substituted benzynes. Treatment of 4,5-dimethylbenzyne (1b) with cyclohexylallene in the presence of the nickel catalyst furnished dihydrophenanthrene derivative 3j in 49% yield. In the same fashion, 1b reacted with n-butylallene and tert-butylallene to give dihydrophenanthrene derivatives 3k and 3l in 46 and 49% yields, respectively

The foregoing results reveal that all the monosubstituted allenes use their internal C–C double bond for the cocyclotrimerization, but for 1,1-disubstituted allenes both C–C double bonds are involved in the reaction. It is not yet clear why the present catalytic reaction strongly favors the participation of the internal double bond of allenes. However, the observation of products **3h'** and **3i'** from the cycloaddition of the respective terminal double bonds of the 1,1-disubstituted allenes is likely due to the steric effect of the two substituents inhibiting the coordination of the internal double bond of the allene to the nickel center.

Based on the above results and known organometallic chemistry, a plausible mechanism is proposed to account for the formation of 9-methylene-9,10-dihydrophenanthrene products (Scheme 2). The catalytic reaction is likely initiated by the reduction of Ni(II) species to Ni(0) species by zinc metal. Selective coordination of the internal C–C double bond of the allene and then benzyne to the nickel center

Table 1 Results of nickel-catalyzed cocyclotrimerization of arynes with allenes^a

Entry	1	2	Product	Yield (%) ^b
1	1 a	2a	Jan	67
2	1 a	2b	3b	58
3	1 a	2c	3c	64
4	1 a	2d	3d	61
5	1 a	2e	3e	64
6	1 a	2f	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	60
7	1 a	2g	~~~~~ 3g	55
8	1 a	2h	3h/3h'	(1 : 1.9) 67
9	1 a	2i	3i/3i/	(2.9 : 1) 69
10	1b	2a	3j	49
11	1b	2e	3k	46
12	1b	2d	31	49
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^{*a*} Unless stated otherwise, all reactions were carried out using benzyne precursor (1) (0.50 mmol), allene (2) (1.0 mmol), NiBr₂(dppe) (10.0 mol%), CsF (2.0 mmol) and Zn (2.75 equiv.) in CH₃CN (2.0 mL) at 80 °C under N₂ for 24 h. ^{*b*} Isolated yields based on benzyne precursor used.

followed by cyclometalation produces nickelacyclopentene intermediate 5.⁷ Coordination of another benzyne and insertion into a Ni(π)–carbon bond gives nickelacycloheptadiene intermediate 6. Subsequent reductive elimination of 6 affords product 3 and regenerates the Ni(0) catalyst. The other way of cyclometalation of 4 gives intermediate 7, but this is less likely due to the fact that 7 has a bulky sp³ carbon attaching the nickel center resulting in strong repulsion between R² and the nickel center and dppe ligand (Fig. 1).

An alternative mechanism involves the coordination of two molecules of benzyne to the nickel center to produce a fivemembered nickelacyclopentadiene intermediate **8** and insertion of the internal C–C double bond of the allene into a Ni(π)–carbon bond to yield intermediate **6**, followed by reductive elimination to give the final product. This mechanism cannot be totally ruled out.

In conclusion, we have developed a novel methodology for the [2+2+2] cocyclotrimerization of arynes with allenes catalyzed by nickel complexes. This is the first report that allenes can undergo



cycloaddition with benzynes in a highly selective fashion to furnish dihydrophenanthrene derivatives in moderate to good yields. Studies to determine the exact mechanistic pathway for this useful reaction are currently underway.

We thank the National Science Council of the Republic of China (NSC-91-2113-M-007-053) for the support of this research.

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